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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/827,529	04/20/2004	Shuhei Yada	1417-460	2966
23117 759 NIXON & VAND		EXAMINER		
901 NORTH GLEBE ROAD, 11TH FLOOR ARLINGTON, VA 22203			PUTTLITZ, KARL J	
			ART UNIT	PAPER NUMBER
			1621	
SHORTENED STATUTORY P	PERIOD OF RESPONSE	MAIL DATE	DELIVER	Y MODE
3 MONT		01/29/2007	PAPER	

Please find below and/or attached an Office communication concerning this application or proceeding.

If NO period for reply is specified above, the maximum statutory period will apply and will expire 6 MONTHS from the mailing date of this communication.

	Application No.	Applicant(s)					
·	10/827,529	YADA ET AL.					
Office Action Summary	Examiner	Art Unit					
	Karl J. Puttlitz	1621					
The MAILING DATE of this communication app	ears on the cover sheet with the c	correspondence address					
Period for Reply	(IO OFT TO EVEIDE AMONITU	(C) OR THIRTY (20) DAVE					
A SHORTENED STATUTORY PERIOD FOR REPLY WHICHEVER IS LONGER, FROM THE MAILING DA - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 16(a). In no event, however, may a reply be tin 17 rill apply and will expire SIX (6) MONTHS from 18 cause the application to become ABANDONE	N. nely filed the mailing date of this communication. D (35 U.S.C. § 133).					
Status							
1) Responsive to communication(s) filed on 17 No.	ovember 2006.						
•—	•						
3) Since this application is in condition for allowar	☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is						
closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213.							
Disposition of Claims							
4)⊠ Claim(s) <u>1,2 and 4-9</u> is/are pending in the application.							
4a) Of the above claim(s) is/are withdrawn from consideration.							
5) Claim(s) is/are allowed.							
6) ☐ Claim(s) is/are rejected.							
7) Claim(s) is/are objected to.	7) Claim(s) is/are objected to.						
8) Claim(s) are subject to restriction and/or	r election requirement.						
Application Papers							
9)☐ The specification is objected to by the Examine	r.						
10) ☐ The drawing(s) filed on is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.							
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).							
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).							
11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.							
Priority under 35 U.S.C. § 119		•					
12)⊠ Acknowledgment is made of a claim for foreign a)⊠ All b)□ Some * c)□ None of:	priority under 35 U.S.C. § 119(a)-(d) or (f).					
1. Certified copies of the priority documents	s have been received.	•					
2. Certified copies of the priority documents have been received in Application No							
3. Copies of the certified copies of the prior	· ·	ed in this National Stage					
application from the International Bureau * See the attached detailed Office action for a list	• • • • • • • • • • • • • • • • • • • •	ed WD					
KARL PUTTLITZ							
		PATENT EXAMINER					
A44b		• • • •					
Attachment(s) 1) Notice of References Cited (PTO-892)	4) 🔲 Interview Summary	γ (PTO-413)					
2) Notice of Draftsperson's Patent Drawing Review (PTO-948)							
3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date	5)	ratent Application					
S. Patent and Trademark Office							

Art Unit: 1621

DETAILED ACTION

Claims 10-13 remain withdrawn. Consequently, claims 1, 2, and 4-9 are subject to reconsideration below:

Receipt is acknowledged of the Priority Document.

The objection to claim 3 is most since this claim has been canceled.

The prior art rejection are maintained and repeated below. Applicant's remarks in connection with this ground of rejection are also addressed.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1, 2 and 4-9 are rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. Patent No. 4,260,821 to Benjamin et al. (Benjamin) in view of EP 1041062 to NIPPON SHOKUBAI CO., LTD. (EP 062).

The rejected claims are drawn to a process for producing (meth)acrylic acid, comprising: contacting a reaction gas containing (meth)acrylic acid obtained by gasphase catalytic oxidation, with an absorbent solvent to prepare a (meth)acrylic acid

solution; and introducing the (meth)acrylic acid solution into a distillation column to purify (meth)acrylic acid, after adjusting a dissolved oxygen concentration in the (meth)acrylic acid solution to be introduced into the distillation column to not less than 12 ppm by weight, the (meth)acrylic acid solution being fed to the distillation column.

The rejected claims are also drawn to those embodiments wherein the (meth)acrylic acid solution to be introduced into the distillation column is mixed with oxygen or an oxygen-containing gas to adjust the dissolved oxygen concentration in the (meth)acrylic acid solution.

The rejected claims are also drawn to those embodiments wherein the (meth)acrylic acid solution to be introduced into the distillation column is mixed with oxygen or an oxygen-containing gas, and then introduced into the distillation column.

The rejected claims are also drawn to those embodiments wherein the (meth)acrylic acid solution to be introduced into the distillation column is mixed with oxygen or an oxygen-containing gas, subjected to a gas-liquid separation, and then introduced into the distillation column.

The rejected claims are also drawn to those embodiments wherein the mixing of the (meth)acrylic acid solution with oxygen or the oxygen-containing gas is performed in a conduit for introducing the (meth)acrylic acid solution into the distillation column, or a static mixer or an orifice disposed in the conduit.

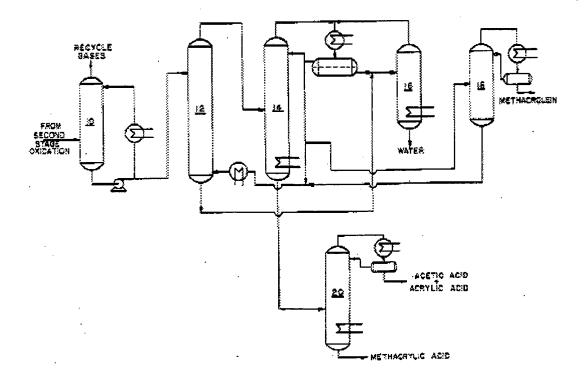
The rejected claims are also drawn to those embodiments wherein the dissolved oxygen concentration in the (meth)acrylic acid solution is adjusted in a facility disposed on an upstream side of the distillation column.

Art Unit: 1621

The rejected claims are also drawn to those embodiments wherein the (meth)acrylic acid solution is in the form of an aqueous solution, the distillation column is an azeotropic dehydration distillation column, and at least a part of a phenol-based polymerization inhibitor is fed to the azeotropic dehydration distillation column from a raw material feed stage thereof or a position higher than the raw material feed stage, and a copper-based polymerization inhibitor is fed to the azeotropic dehydration column from a position lower than the raw material feed stage.

The rejected claims are also drawn to those embodiments wherein the azeotropic dehydration column is any of a perforated plate column, a packed column and a combination of a perforated plate column and a packed column.

With regard to the above embodiments, Benjamin teaches a methacrylic acid recovery process in conjunction with the following figure 1:



Art Unit: 1621

wherein the cooling and condensation of reactor effluent gases in quench tower 10. Residual gases are separated and recycled to the second-stage reactor (i.e., a gas liquid separation, see claim 4). Also, the tower 10 is equipped with a valve (see claim 6). Extraction of crude methacrylic acid with a suitable solvent is carried out in column 12. Purification of the acid includes separation of the solvent from crude methacrylic acid (including acetic acid) by distillation in column 14 with recirculation of the solvent to the extraction step. Crude methacrylic acid is separated into a pure methacrylic acid product and a by-product stream containing acetic acid and acrylic acid as the principal constituents in column 20. Purification of the acetic acid may be carried out in subsequent distillations, not shown.

Although most of the water in the reactor effluent is rejected during the solvent extraction, a certain amount of water is carried into the solvent recovery column 14 and is separated from the solvent in the overhead facilities. The combined water from the extraction column 12 and solvent recovery column 14 is overhead is stripped of dissolved solvent in column 16 and rejected, while the recovered solvent is returned to the solvent recovery column (14) overhead equipment. A portion of the unreacted methacrolein tends to accumulate in the solvent and recovery by distillation of a slip stream of solvent is generally economically justified. Such a column is shown as 18 in FIG. 1.

The patent teaches that fouling and/or plugging of the distillation columns, and particularly their reboilers, is a common problem when processing solvent rich in methacrylic acid and acetic acid. The nature of the fouling materials is not fully known,

Art Unit: 1621

however, it has been found that by use of conventional inhibitors, particularly hydroquinone, fouling may be greatly reduced, but only when used in conjunction with oxygen injection at a rate above a predetermined threshold value which has been found necessary to prevent serious fouling of the equipment.

Benjamin teaches that inhibitors which may be used in conjunction with the present invention include those known in the art such as aromatic phenols, aromatic amines, and quinones. Hydroquinone and the commercially available A-30, 2,4 dimethyl 6 t-butyl phenol available from DuPont, along with Topanol A, 2,4 dimethyl 6 t-butyl phenol available from ICI United States, Inc., have been found to be useful. The amounts of inhibitor(s) will be generally similar to those known in the art, typically between 100 and 1000 ppm (wt). See columns 3 and 4.

In particular, Benjamin teaches that the appearance of polymers and/or other high-boiling materials as solids during the recovery of methacrylic acid produced by oxidation of methacrolein can be minimized by the use of known inhibitors such as hydroquinone and the like, and the introduction of molecular O_2 , in the form of air or enriched air, at a rate above a threshold value, defined as the amount of oxygen above which no significant reduction of the rate of appearance of solids occurs. The threshold value is primarily affected by the temperature of the boiling liquids where solid polymers are expected to appear. At an operating temperature of about 120°C the threshold value is at about 0.1 SLH O_2 /100 gms of liquid. See column 2, lines 43-64, and claim 1.

The difference between the process covered in the rejected claims and the process disclosed in Benjamin, is that Benjamin fails to expressly disclose that the

Art Unit: 1621

(meth)acrylic acid solution is fed to the distillation column after adjusting a dissolved oxygen concentration in the (meth)acrylic acid solution, or more specifically, a conduit for introducing the (meth)acrylic acid solution into the distillation column, or a static mixer or an orifice disposed in the conduit, or facility upstream from the distillation column. Also, Benjamin fails to teach copper-based polymerization inhibitors (claim 8), or the specific columns required by the claims, e.g., an azeotropic distillation column. It is for this proposition, however, that the examiner joins EP 062. In this regard, EP 062 teaches a process for producing (meth)acrylic acid. In particular, the patent teaches that a polymerization inhibitor adding to a distillation no particular restriction. This polymerization inhibitor may be directly fed into the distillation column or it may be dissolved in a feed solution, a reflux liquid, or other solvent and then fed in the form of the resultant solution to the distillation column via the feed line. See paragraph [0027]. Copper-based polymerization inhibitors are taught in paragraph [0026]. EP 062 also teaches that examples of the distillation column may be cited a solvent separation column, an azeotropic separation column, an acetic acid separation column, and a high boiling component separation column. See paragraph [0028].

Accordingly, EP 062 demonstrates that adding a solids/polymerization inhibitor before distillation of methacrylic acid in an azeotropic distillation column, and moreover, with a copper based polymerization inhibitor, is routine in the art, and thus, well within the motivation of those of ordinary skill. Accordingly, those of ordinary skill would have been motivated to modify the disclosure of Benjamin to include a step of adding a solids/polymerization inhibitor before distillation of methacrylic acid in the required

Art Unit: 1621

columns, and additionally, with a copper based polymerization inhibitor, since EP 062 demonstrates that such steps are fundamental in methacrylic acid recovery systems. Therefore, the rejected claims are prima facie obvious since the combination of Benjamin and EP 062 teaches, or suggests, the elements of these claims with a reasonable expectation of success.

Page 8

The instant action is a non-final rejection to clarify the oxygen concentration requirements of the instant claims:

Applicant argues that the applied references, either alone or in combination, do not teach or suggest the required dissolved oxygen concentration. However, given the fact that water is used as an absorbent, and given the fact that water is only a portion of the material (besides, methacrylic acid, side-products, and polymerization inhibitor) entering the distillation column of the references, those of ordinary skill would expect that the dissolved oxygen concentration of the references would be within the range required by the claims. Indeed, the record is absent any objective evidence to the contrary, see MPEP (""[T]he PTO can require an applicant to prove that the prior art products do not necessarily or inherently possess the characteristics of his [or her] claimed product. Whether the rejection is based on 'inherency' under 35 U.S.C. 102, on 'prima facie obviousness' under 35 U.S.C. 103, jointly or alternatively, the burden of proof is the same...[footnote omitted]." The burden of proof is similar to that required with respect to product-by-process claims. In re Fitzgerald, 619 F.2d 67, 70, 205 USPQ

Art Unit: 1621

594, 596 (CCPA 1980) (quoting In re Best, 562 F.2d 1252, 1255, 195 USPQ 430, 433-34 (CCPA 1977)).").

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Karl J. Puttlitz whose telephone number is (571) 272-0645. The examiner can normally be reached on Monday to Friday from 9 a.m. to 5 p.m.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Thurman K. Page, can be reached at telephone number (571) 272-0602. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

> KARL PUTTLITZ PATENT EXAMINER